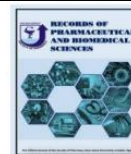




RECORDS OF PHARMACEUTICAL AND BIOMEDICAL SCIENCES



Chemistry and biological activities of *Cichorium endivia*: A mini review

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Abstract

Cichorium endivia L. is a wild edible plant and widely distributed in the Mediterranean region. Traditionally, it had been used to relief of mild digestive disorders symptoms such as flatulence, feeling of abdominal fullness, and slow digestion, and loss of appetite. It is a bitter-leaved vegetable with simple, alternate and sessile leaves forming an inflorescence. Recently, *C. endivia* gained attention because of its therapeutic and biological activities such as antimicrobial, hepatoprotective, antioxidant and antiproliferative. In this review, we summarize the most important chemical constituents and biological activities *C. endivia*.

Keywords: *Cichorium endivia*; chemical constituents; biological activities.

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1. Introduction:

Cichorium endivia L. subsp. *Pumilum* Jacq., (Arabic name: Shikorya or Sereis), commonly known as chicory, is a wild edible plant and belongs to family Asteraceae. It is a bitter-leaved vegetable with simple, alternate and sessile leaves forming an inflorescence. Additionally, the leaves are slightly pubescent, pale to dark green, sometimes reddish along midrib and the flowers are blue (Figure 1) (Aisa et al., 2020).

It is widely distributed in the Mediterranean region and cultivated in many countries in Asia, Europe and North America (Khalil & Kamel, 2015; Amer, 2018). *C. endivia* L. is valuable nutritionally with a high content of vitamin C, minerals and dietary fibres (Kopeck, 1998; Koudela & Petřiková, 2007). So that, it sometimes used as a vegetable in salad and in the traditional folklore was used to relief of mild digestive disorders symptoms such as flatulence, feeling of abdominal fullness, and slow digestion, and loss of appetite (Masoud et al., 2018). In addition, it is famous among the Egyptian

farmers and is preferred to be eaten with cheese as a common Egyptian meal (Abou-Zeid, 2015; Masoud et al., 2018). Furthermore, leaves decoction was used for poisoning, bacterial infection, rheumatism (Azaizeh et al., 2006) and diabetes (Al Khateeb et al., 2012).

C. endivia L. showed many biological activities such as antimicrobial (Amer, 2018), hepatoprotective, (Chen et al., 2011), antioxidant (Papetti et al., 2002) and antiproliferative (Wang et al., 2012; Alshehri & Elsayed, 2012).

Phytochemical investigation of *C. endivia* L. revealed presence of many bioactive compounds, such as flavonoids (Saleh et al., 1975; Mascherpa et al., 2012; Hegazy et al., 2015) coumarins (Khalil and Kamel, 2015), Sesquiterpenes and their glycosides (Seto et al., 1988; Kisiel and Michalska, 2003), phenolic acids (Kisiel and Michalska, 2006; Papetti et al., 2008) and nitrogenous compounds (Chen et al., 2011; Wang et al., 2012; Aisa et al., 2020).

2. Phytochemical constituents reported in *Cichorium endivia* L.:

A review on the phytochemical compounds and the pharmacological activities of *Cichorium endivia* L. was done and it revealed that the chemical compounds isolated and identified from the *C.*

endivia belong to various chemical classes including sterols, triterpenes, sesquiterpenes, flavonoids, phenolic and organic acids, phenolic glycosides, coumarins, nitrogen containing compounds along with many other miscellaneous compounds as summarized in the tables (1-7).

Table 1: Sterols and triterpenoids:

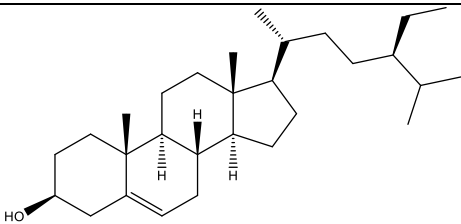
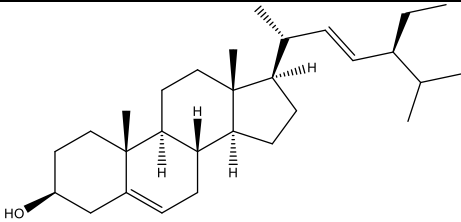
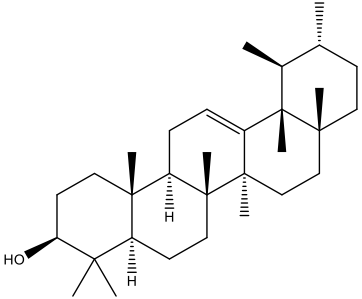
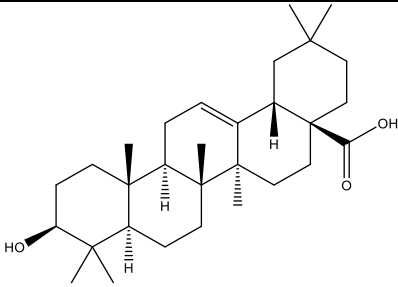
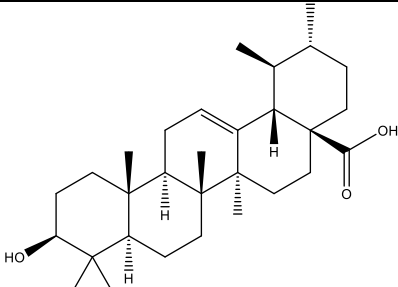
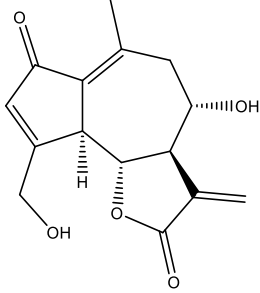
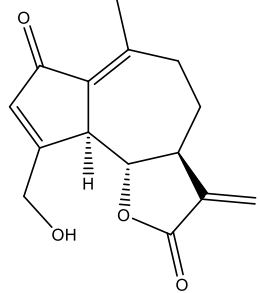
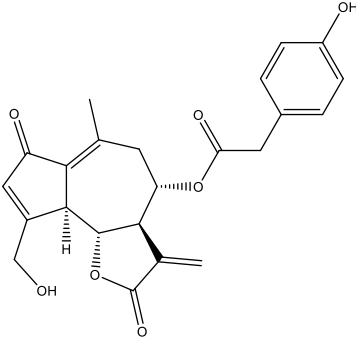
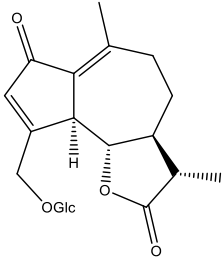
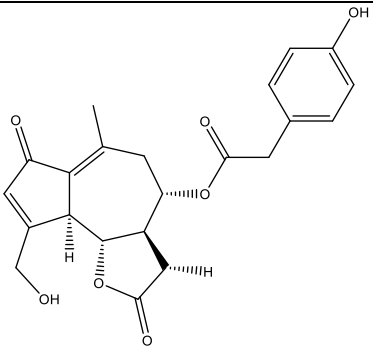
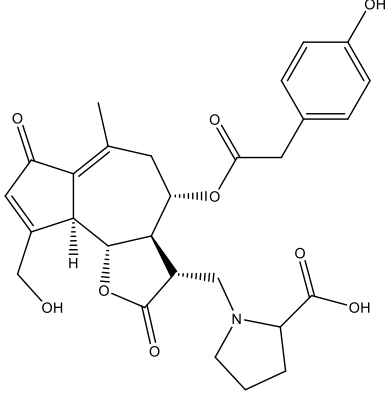
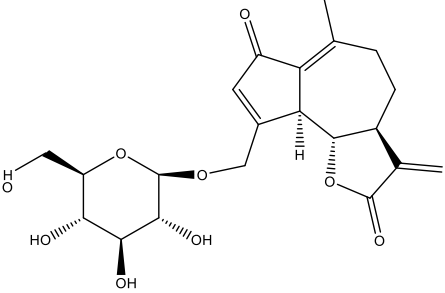
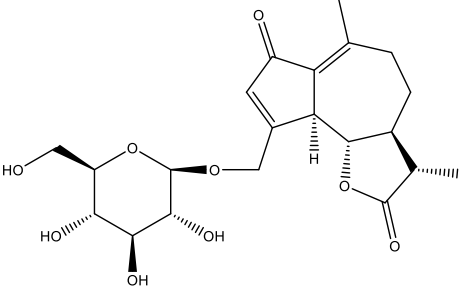
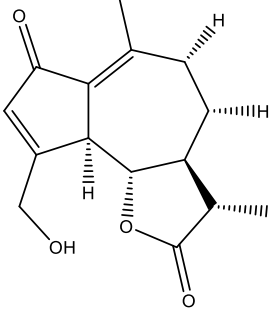
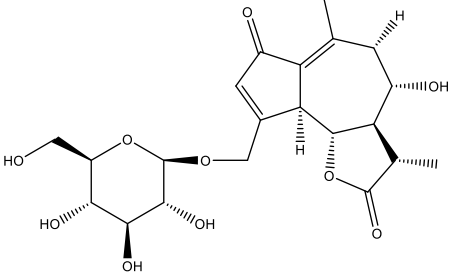
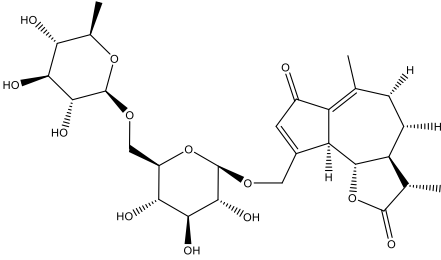
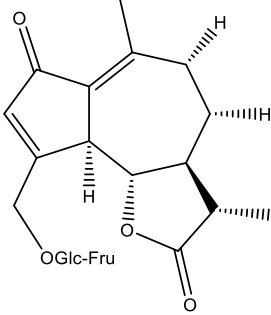
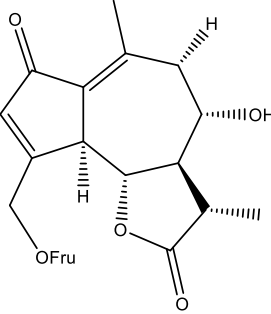
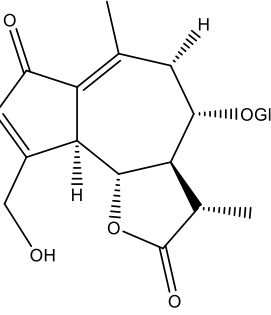
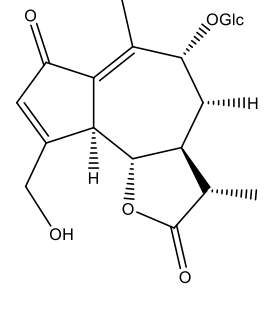
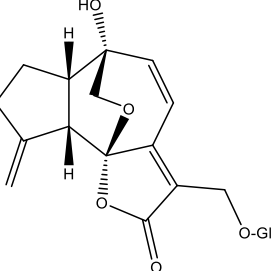
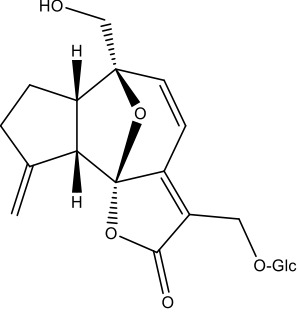
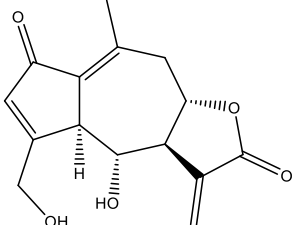
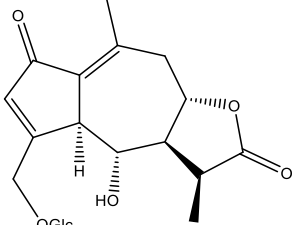
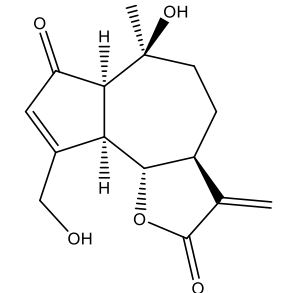
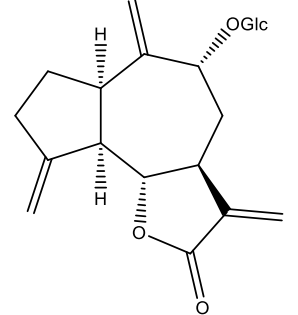
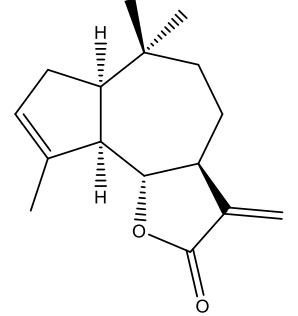
Compound Name	Compound Structure	Reference
β -Setosterol		(Hegazy <i>et al.</i> , 2015)
Stigmasterol		(Hegazy <i>et al.</i> , 2015)
α -Amyrin		(Hegazy <i>et al.</i> , 2015)
Oleanolic acid		(Hegazy <i>et al.</i> , 2015)
Ursolic acid		(Hegazy <i>et al.</i> , 2015)

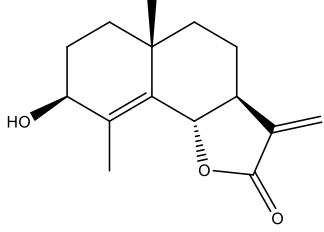
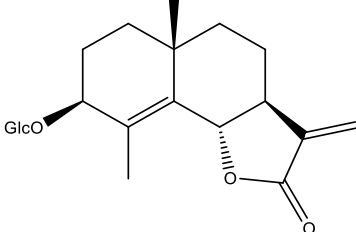
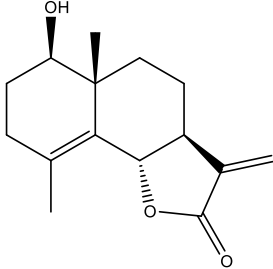
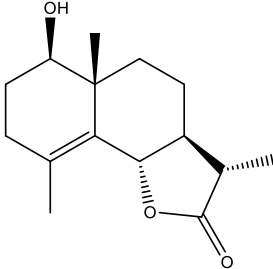
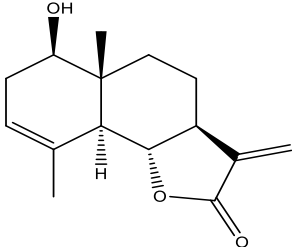
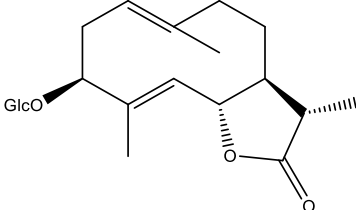
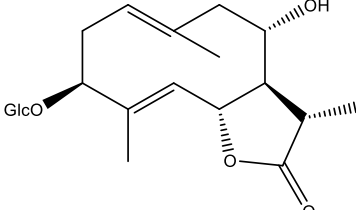
Table 2: Sesquiterpenes:

Compound Name	Compound Structure	Reference
Lactucin		(Seto <i>et al.</i> , 1988)
8-deoxylactucin		(Seto <i>et al.</i> , 1988)
Lactucopicrin		(Seto <i>et al.</i> , 1988)
11 β , 13- dihydrolactucin		(Seto <i>et al.</i> , 1988)
11 β , 13- dihydrolactucopicrin		(Kisiel and Michalska, 2006)

<p>11 β,13-Dihydro-13-prolyl-lactucopicrin</p>	 <p>The structure shows a complex polycyclic lactone core. It features a seven-membered ring fused to a five-membered ring, which is further fused to a six-membered ring. A prolyl group (a five-membered pyrrolidine ring with a carboxylic acid group) is attached to the six-membered ring. A hydroxyl group is attached to the seven-membered ring. A side chain containing a carboxylic acid group and a hydroxyl group is attached to the five-membered ring.</p>	<p>(Warashina and Miyase, 2008)</p>
<p>Crepidiaside A</p>	 <p>The structure shows a complex polycyclic lactone core. It features a seven-membered ring fused to a five-membered ring, which is further fused to a six-membered ring. A glucose molecule is attached to the six-membered ring via an oxygen atom. The glucose molecule has hydroxyl groups at the 2, 3, and 6 positions. A methyl group is attached to the seven-membered ring. A side chain containing a carboxylic acid group and a hydroxyl group is attached to the five-membered ring.</p>	<p>(Seto <i>et al.</i>, 1988)</p>
<p>Crepidiaside B</p>	 <p>The structure shows a complex polycyclic lactone core. It features a seven-membered ring fused to a five-membered ring, which is further fused to a six-membered ring. A glucose molecule is attached to the six-membered ring via an oxygen atom. The glucose molecule has hydroxyl groups at the 2, 3, and 6 positions. A methyl group is attached to the seven-membered ring. A side chain containing a carboxylic acid group and a hydroxyl group is attached to the five-membered ring.</p>	<p>(Seto <i>et al.</i>, 1988)</p>
<p>Jacquinelin</p>	 <p>The structure shows a complex polycyclic lactone core. It features a seven-membered ring fused to a five-membered ring, which is further fused to a six-membered ring. A hydroxyl group is attached to the seven-membered ring. A side chain containing a carboxylic acid group and a hydroxyl group is attached to the five-membered ring.</p>	<p>(Warashina and Miyase, 2008)</p>
<p>Cichorioside B</p>	 <p>The structure shows a complex polycyclic lactone core. It features a seven-membered ring fused to a five-membered ring, which is further fused to a six-membered ring. A glucose molecule is attached to the six-membered ring via an oxygen atom. The glucose molecule has hydroxyl groups at the 2, 3, and 6 positions. A methyl group is attached to the seven-membered ring. A side chain containing a carboxylic acid group and a hydroxyl group is attached to the five-membered ring.</p>	<p>(Warashina and Miyase, 2008)</p>

Cichorioside D	 <p>The structure of Cichorioside D consists of a central bicyclic core with a lactone ring fused to a seven-membered ring. It is substituted with a methyl group, a hydroxyl group, and a side chain containing a double bond and a methyl group. A glucose unit is attached to the side chain via an ether linkage, and a fructose unit is attached to the lactone ring via an ether linkage.</p>	(Warashina and Miyase, 2008)
Cichorioside E	 <p>The structure of Cichorioside E features the same bicyclic core as Cichorioside D. It has a methyl group, a hydroxyl group, and a side chain with a double bond and a methyl group. A glucose-fructose unit (OGlc-Fru) is attached to the side chain via an ether linkage.</p>	(Warashina and Miyase, 2008)
Cichorioside F	 <p>The structure of Cichorioside F is similar to Cichorioside E, but the glucose-fructose unit is replaced by a fructose unit (OFru) attached to the side chain via an ether linkage.</p>	(Warashina and Miyase, 2008)
Cichorioside G	 <p>The structure of Cichorioside G has the same bicyclic core. It features a methyl group, a hydroxyl group, and a side chain with a double bond and a methyl group. A glucose unit (OGlc) is attached to the side chain via an ether linkage, and a hydroxyl group (OH) is attached to the lactone ring.</p>	(Warashina and Miyase, 2008)
Cichorioside H	 <p>The structure of Cichorioside H is similar to Cichorioside G, but the hydroxyl group on the lactone ring is replaced by a glucose unit (OGlc) attached via an ether linkage.</p>	(Warashina and Miyase, 2008)
Cichorioside J	 <p>The structure of Cichorioside J is a bicyclic core with a lactone ring. It has a methyl group, a hydroxyl group, and a side chain with a double bond and a methyl group. A glucose unit (O-Glc) is attached to the side chain via an ether linkage.</p>	(Warashina and Miyase, 2008)

<p>Cichorioside K</p>		<p>(Warashina and Miyase, 2008)</p>
<p>Intybulide</p>		<p>(Kisiel and Michalska, 2006)</p>
<p>Cichorioside I</p>		<p>(Warashina and Miyase, 2008)</p>
<p>Hieracin II</p>		<p>(Kisiel and Michalska, 2006)</p>
<p>Ixerisoside D</p>		<p>(Kisiel and Michalska, 2006)</p>
<p>Macroclinside G</p>		<p>(Kisiel and Michalska, 2006)</p>

Cichoriolide A		(Seto <i>et al.</i> , 1988)
Cichorioside A		(Seto <i>et al.</i> , 1988)
Magnolialide		(Kisiel and Michalska, 2006)
Artesin		(Kisiel and Michalska, 2003)
Santamarine		(Kisiel and Michalska, 2003)
Sonchuside A		(Seto <i>et al.</i> , 1988)
Cichorioside C		(Warashina and Miyase, 2008)

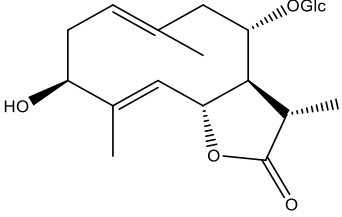
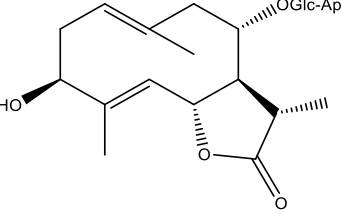
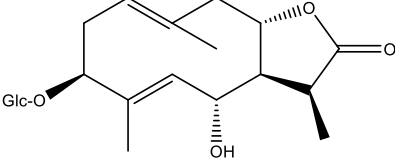
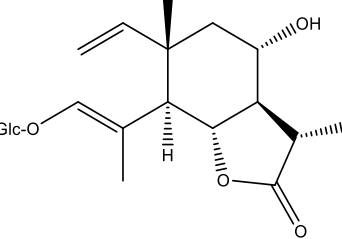
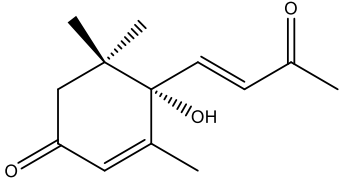
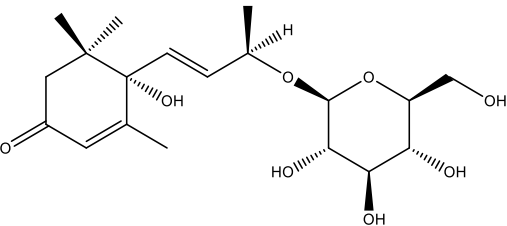
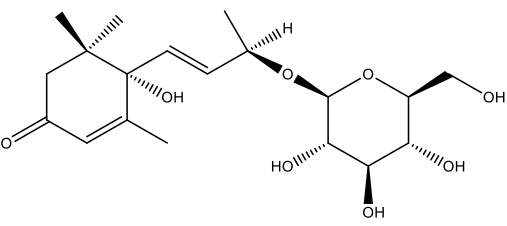
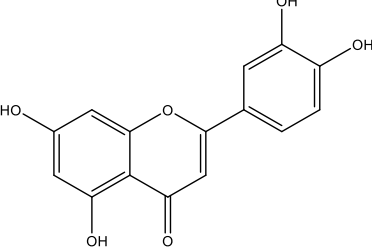
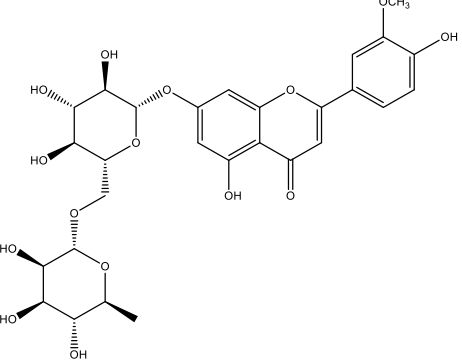
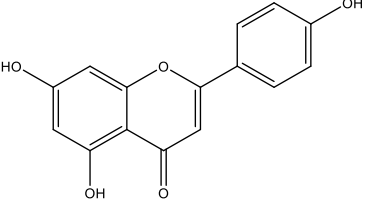
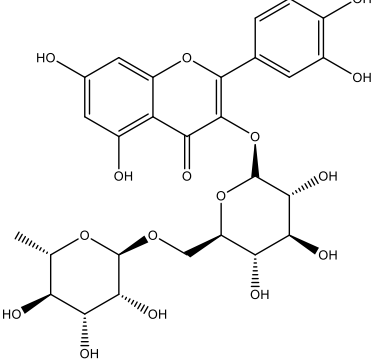
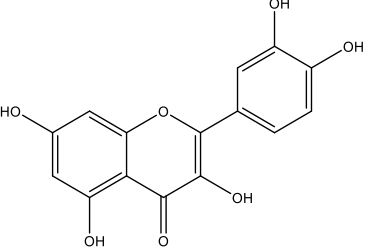
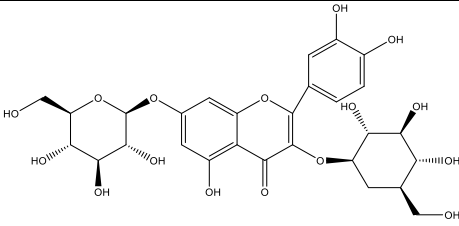
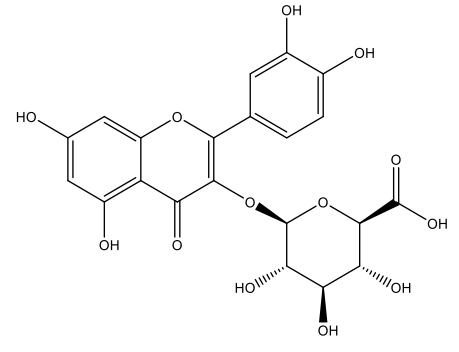
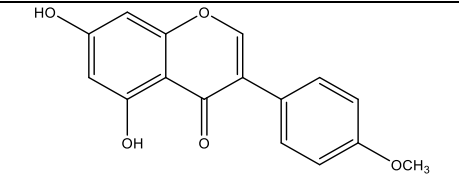
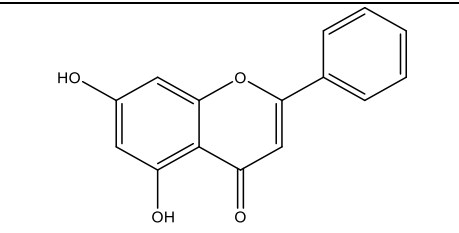
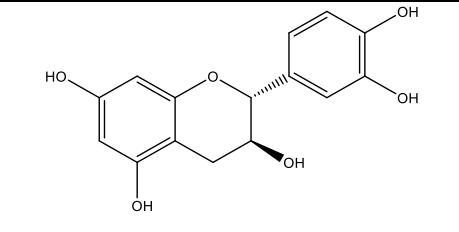
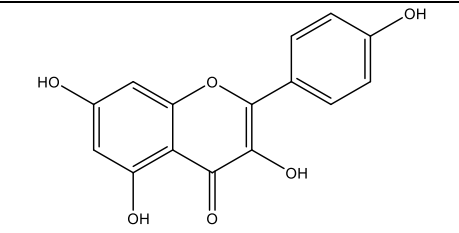
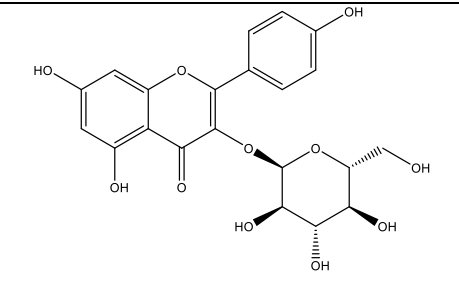
<p>Hypochoeroside A</p>		<p>(Warashina and Miyase, 2008)</p>
<p>Cichorioside L</p>		<p>(Warashina and Miyase, 2008)</p>
<p>Cichorioside M</p>		<p>(Warashina and Miyase, 2008)</p>
<p>Cichorioside N</p>		<p>(Warashina and Miyase, 2008)</p>
<p>(6S, 7E)-6-hydroxy-4,7-megastigmadien-3,9-dione (S (+)-dehydrovomifoliol)</p>		<p>(Kisiel <i>et al.</i>, 2004)</p>
<p>(6S, 7E, 9R)-6,9-dihydroxy-4,7-megastigmadien-3-one 9-O-b-glucopyranoside (roseoside)</p>		<p>(Kisiel <i>et al.</i>, 2004)</p>
<p>(6S, 7E, 9S)- 6,9-dihydroxy-4,7-megastigmadien-3-one 9-O-b-glucopyranoside (diastereomer)</p>		<p>(Kisiel <i>et al.</i>, 2004)</p>

Table 3: Flavonoids:

Compound Name	Compound Structure	Reference
Luteolin		(El-Shafey and AbdElgawad, 2012)
Luteolin-3'-methoxy-7-rutinoside		(Hegazy et al., 2015)
Apigenin		(Hegazy et al., 2015)
Rutin		(Hegazy et al., 2015)
Quercetin		(Hegazy et al., 2015)

<p>Quercetin-3,7-di-O-glucoside</p>	 <p>The structure shows a quercetin aglycone core with two glucose units attached at the 3 and 7 positions of the flavanone ring. The glucose units are in their cyclic pyranose form, with hydroxyl groups at various positions and a hydroxymethyl group at the C6 position.</p>	<p>(Mascherpa <i>et al.</i>, 2012)</p>
<p>Quercetin-monoglucuronide</p>	 <p>The structure shows a quercetin aglycone core with a single glucuronic acid unit attached at the 3 position of the flavanone ring. The glucuronic acid is in its cyclic form, featuring a carboxylic acid group at the C5 position.</p>	<p>(Mascherpa <i>et al.</i>, 2012)</p>
<p>5,7-dihydroxy 4'-methoxy isoflavone</p>	 <p>The structure shows an isoflavone core with hydroxyl groups at the 5 and 7 positions of the A-ring and a methoxy group at the 4' position of the B-ring.</p>	<p>(Hegazy <i>et al.</i>, 2015)</p>
<p>Chrysin</p>	 <p>The structure shows a flavanone core with hydroxyl groups at the 5 and 7 positions of the A-ring and a phenyl group at the 2 position of the C-ring.</p>	<p>(Hegazy <i>et al.</i>, 2015)</p>
<p>Catechin</p>	 <p>The structure shows a flavanone core with hydroxyl groups at the 5 and 7 positions of the A-ring and a catechol group at the 2 position of the C-ring.</p>	<p>(Hegazy <i>et al.</i>, 2015)</p>
<p>Kaempferol</p>	 <p>The structure shows a flavanone core with hydroxyl groups at the 5 and 7 positions of the A-ring and a p-hydroxyphenyl group at the 2 position of the C-ring.</p>	<p>(Chen <i>et al.</i>, 2011)</p>
<p>Kaempferol-3-O-β-D-glucoside (Astragalin)</p>	 <p>The structure shows a kaempferol aglycone core with a β-D-glucose unit attached at the 3 position of the flavanone ring. The glucose is in its cyclic form with hydroxyl groups at the 2, 3, and 6 positions.</p>	<p>(Chen <i>et al.</i>, 2011)</p>

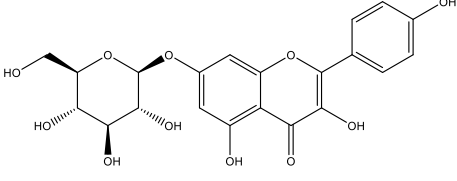
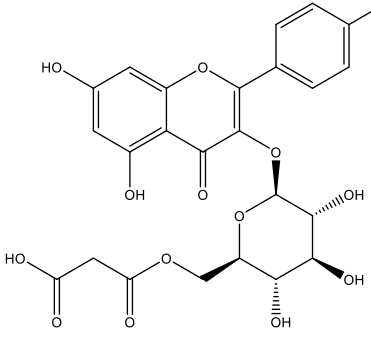
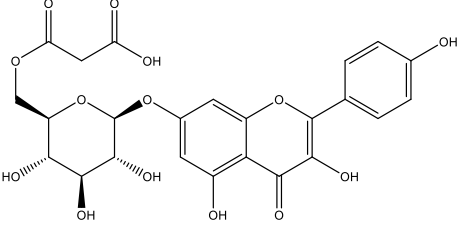
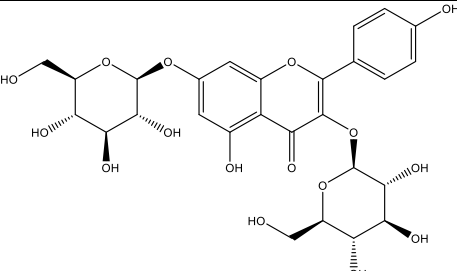
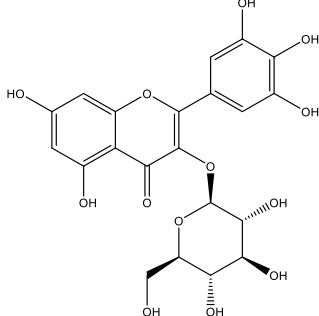
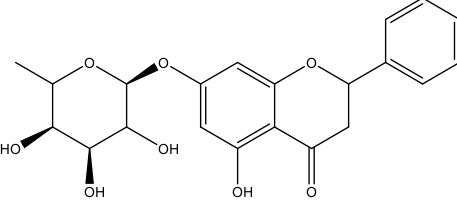
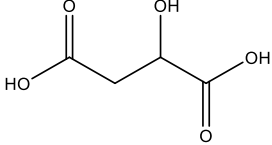
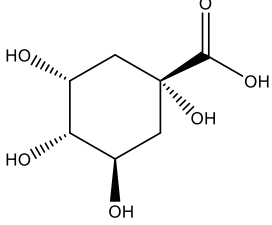
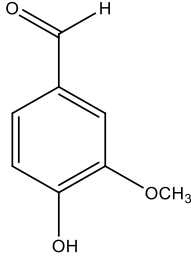
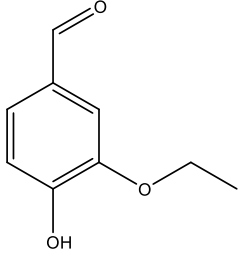
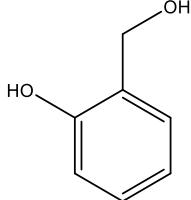
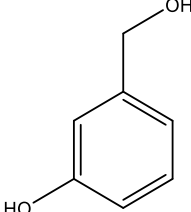
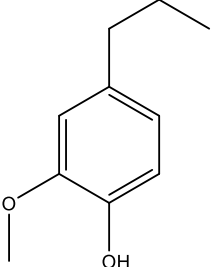
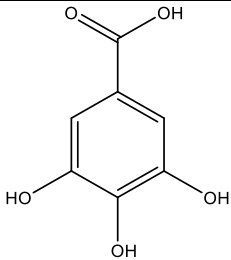
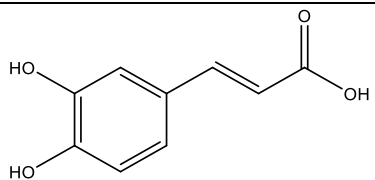
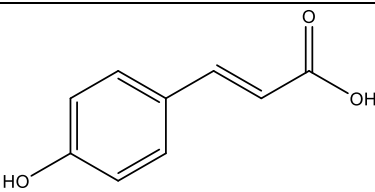
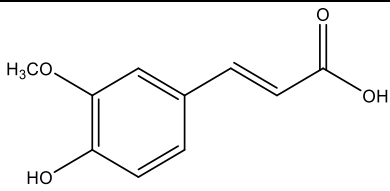
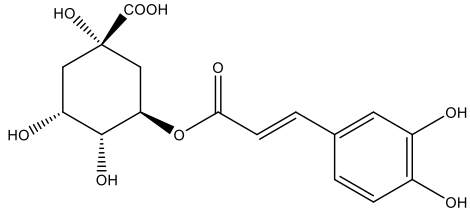
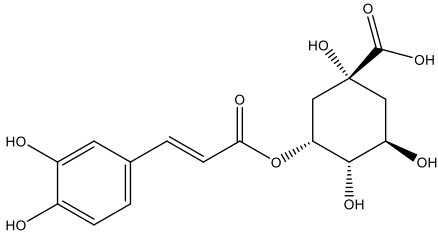
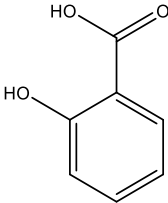
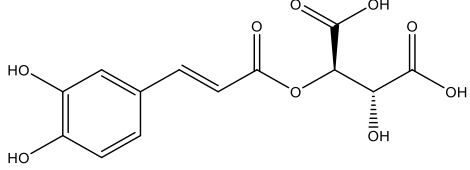
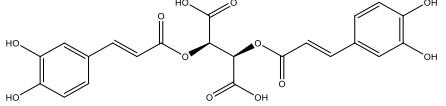
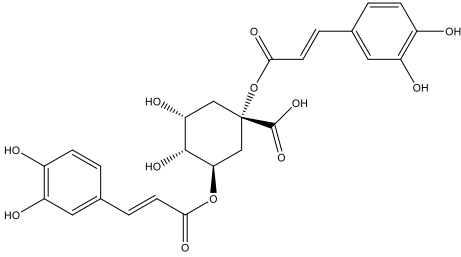
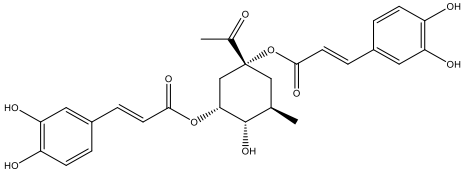
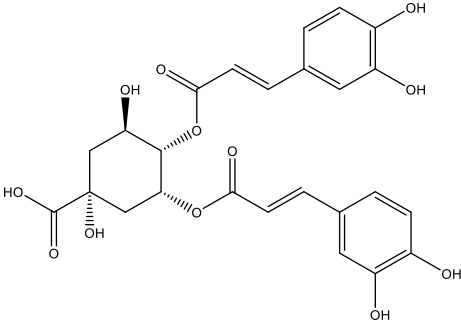
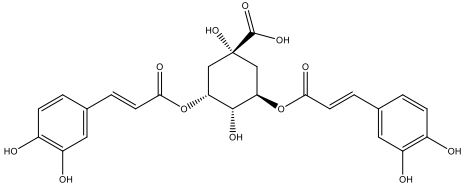
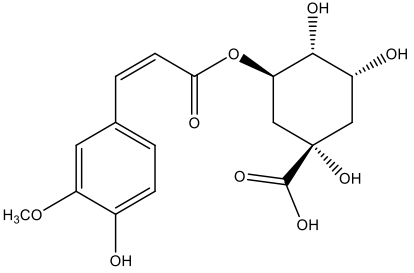
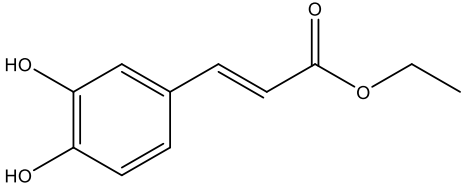
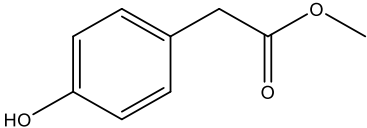
Kaempferol-7-O-glucoside		(Mascherpa <i>et al.</i> , 2012)
Kaempferol-3-O-(6''-malonyl)-glucoside		(Mascherpa <i>et al.</i> , 2012)
Kaempferol-7-O-(6''-malonyl)-glucoside		(Mascherpa <i>et al.</i> , 2012)
Kaempferol-O-diglucoside		(Mascherpa <i>et al.</i> , 2012)
Myricetin-3-O-glucoside		(Mascherpa <i>et al.</i> , 2012)
Pinocembrin-O-rhamnoside		(Mascherpa <i>et al.</i> , 2012)

Table 4: Organic and phenolic acids and their derivatives:

1. Compound Name	Compound Structure	Reference
Malic acid		(Mascherpa <i>et al.</i> , 2012)
Quinic acid		(Mascherpa <i>et al.</i> , 2012)
Vanillin		(Hegazy <i>et al.</i> , 2015)
Ethyl vanillin		(Enk <i>et al.</i> , 2004)
Salicyl alcohol		(Enk <i>et al.</i> , 2004)
3-hydroxy benzenemethanol		(Enk <i>et al.</i> , 2004)
Dihydroeugenol		(Enk <i>et al.</i> , 2004)

Gallic acid		(Hegazy <i>et al.</i> , 2015)
Caffeic acid		(Mascherpa <i>et al.</i> , 2012)
<i>p</i> -Coumaric acid		(Hegazy <i>et al.</i> , 2015)
Ferulic acid		(Hegazy <i>et al.</i> , 2015)
Chlorogenic acid (3-caffeoylquinic)		(Hegazy <i>et al.</i> , 2015)
5-caffeoylquinic (5-CQA) (Neochlorogenic acid)		(Mikropoulou <i>et al.</i> , 2018)
Salicylic acid		(Hegazy <i>et al.</i> , 2015)
trans-caftaric acid		(Mascherpa <i>et al.</i> , 2012)

<p>Chicoric acid</p>		<p>(Mascherpa <i>et al.</i>, 2012)</p>
<p>1,3-dicaffeoylquinic acid</p>		<p>(Mascherpa <i>et al.</i>, 2012)</p>
<p>1,5-di-O-caffeoylquinic acid</p>		<p>(Singab <i>et al.</i>, 2010)</p>
<p>3,4-dicaffeoylquinic acid</p>		<p>(Papetti <i>et al.</i>, 2008)</p>
<p>3,5-Di-O-caffeoylquinic acid</p>		<p>(Papetti <i>et al.</i>, 2008)</p>
<p>5-feruloylquinic acid</p>		<p>(Mascherpa <i>et al.</i>, 2012)</p>
<p>Ethyl trans-caffeate</p>		<p>(Kisiel and Michalska, 2006)</p>
<p>Methyl p-hydroxyphenylacetate</p>		<p>(Kisiel and Michalska, 2006)</p>

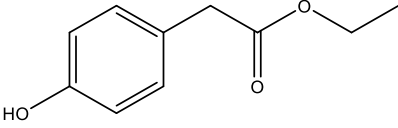
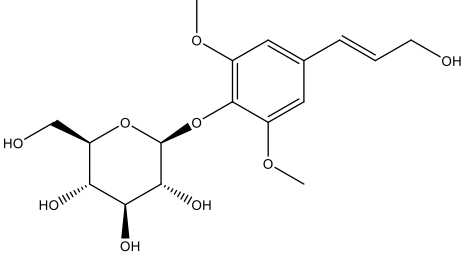
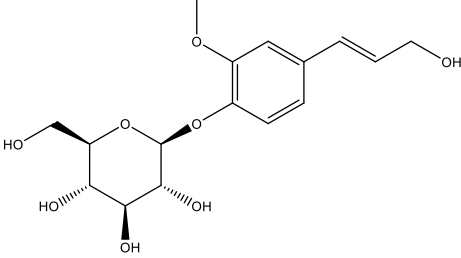
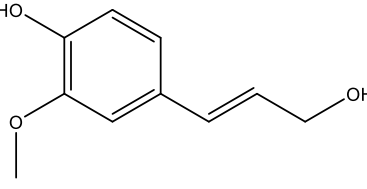
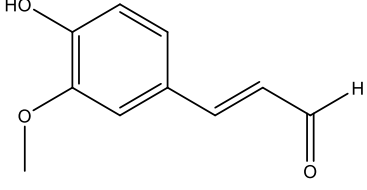
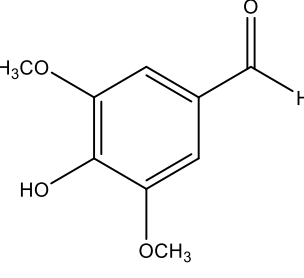
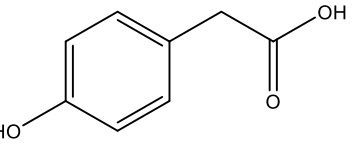
Ethyl p-hydroxyphenylacetate		(Kisiel and Michalska, 2006)
Syringin		(Kisiel and Michalska, 2003)
Coniferin		(Kisiel and Michalska, 2003)
Coniferyl alcohol		(Kisiel and Michalska, 2003)
Coniferyl aldehyde		(Kisiel and Michalska, 2003)
Syringaldehyde		(Kisiel and Michalska, 2003)
4-Hydroxy phenyl acetic acid		(Khalil and Kamel, 2015)

Table 5: Coumarins:

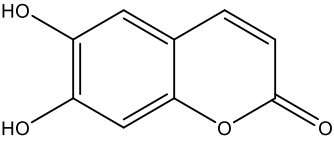
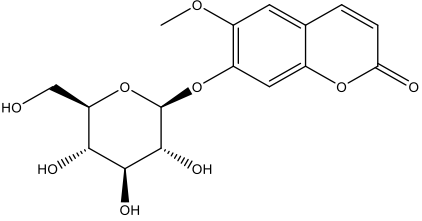
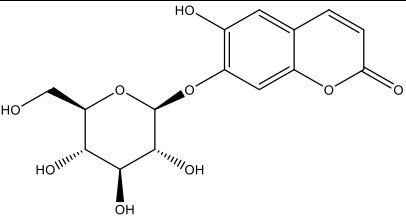
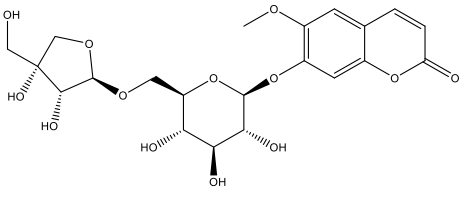
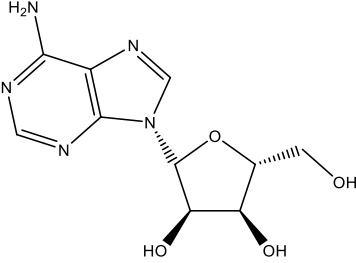
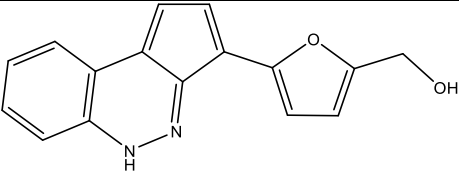
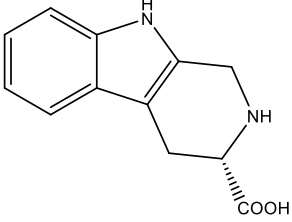
Compound Name	Compound Structure	Reference
Aesculetin		(Singab <i>et al.</i> , 2010)
Scopolin		(Khalil and Kamel, 2015)
Cichoriin		(Khalil and Kamel, 2015)
Hymexelsin		(Khalil and Kamel, 2015)

Table 6: Nitrogen containing compounds:

Compound Name	Compound Structure	Reference
Adenosine		(Chen <i>et al.</i> , 2011)
2-furanmethanol-(5'→11)-1,3-cyclopentadiene-[5,4-c]-1H-cinnoline		(Chen <i>et al.</i> , 2011)
(3S)-1,2,3,4-tetrahydro-β-carboline-3-carboxylic acid		(Wang <i>et al.</i> , 2012)

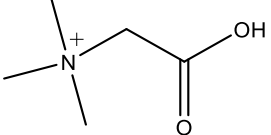
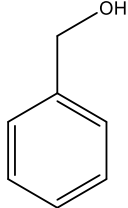
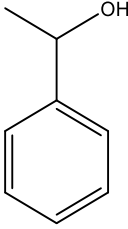
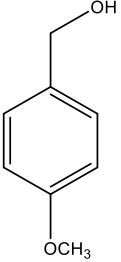
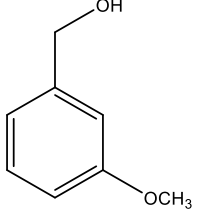
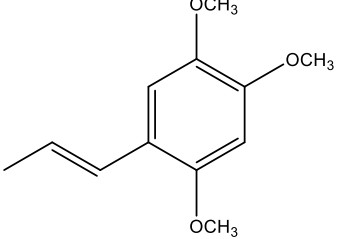
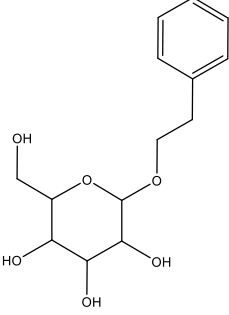
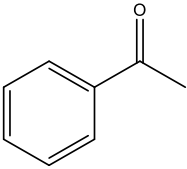
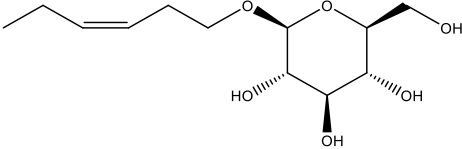
Betaine		(Aisa <i>et al.</i> , 2020)
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Table 7: Other miscellaneous compounds :

Compound Name	Compound Structure	Reference
Benzyl alcohol		(Enk <i>et al.</i> , 2004)
Phenylethyl alcohol		(Enk <i>et al.</i> , 2004)
<i>p</i> -anisyl alcohol		(Enk <i>et al.</i> , 2004)
<i>m</i> -anisyl alcohol		(Enk <i>et al.</i> , 2004)
<i>cis</i> -asaron		(Enk <i>et al.</i> , 2004)

2-phenylethyl- β -D-glucopyranoside		(Chen <i>et al.</i> , 2011)
Acetophenone		(Schmidt and Schreier, 1986)
z-3-Hexenyl- β -D-glucoside		(Khalil and Kamel, 2015)

3. Biological activities reported in *Cichorium endivia L.*:

Studies on *Cichorium endivia L.* revealed that it possesses a wide range of biological activities

including anti-oxidant, antimicrobial, insecticidal, anthelmintic anti-Blastocystis, anti-diabetic antihyperglycemic, antihyperlipidemic, anti-inflammatory, antiangiogenic and cytotoxic activities as summarized in Table 8.

Table 8: Biological activities reported in *Cichorium endivia L.*:

I- Anticancer activity		
Active Compounds / extracts	Details	Reference
3S-1,2,3,4-tetrahydro β -carboline-3-carboxylic acid	Remarkable <i>in-vitro</i> cytotoxicity against HCT-8 and HepG2 cell-lines.	(Wang <i>et al.</i> , 2012)
Hydro-ethanolic extract	Significant <i>in-vivo</i> protective effect of sun light-activated extract against the dimethylbenz[a]anthracene (DMBA) induced benign breast tumors to female rats by reducing the lobular hyperplasia and fibroadenoma induced in the mammary glands.	(Al-Akhras <i>et al.</i> , 2012)
3S-1,2,3,4-tetrahydro β -carboline-3-carboxylic acid	Significant <i>in-vivo</i> anti-proliferative effect on human Colorectal Cancer cell line HCT-8, and induction of apoptosis of HCT-8 cells via the suppression of NF- κ B signaling pathway in a dose-dependent manner.	(Wang <i>et al.</i> , 2012)
Methanolic root extract	Polyphenolic extract from roots using methanol showed significant <i>in-vitro</i> cytotoxic activity against breast cancer	Alshehri and Elsayed, 2012)

	cell line (MCF7) with remarkable changes in the gene expression for the DNA cancer markers.	
aqueous decoction	Remarkable <i>in-vitro</i> cytotoxic activity in C5N cells which represent an immortalized highly differentiated non-tumorigenic cell line	(Mikropoulou <i>et al.</i> , 2018)
II- Antimicrobial activity		
Methanolic extract	Methanolic extract at 24°C showed antibacterial activity against both gram negative bacteria: <i>Klebsiella pneumoniae</i> (ATCC 10031) and <i>Pseudomonas aeruginosa</i> (ATCC 27853) as well as gram positive bacteria: <i>Bacillus cereus</i> (ATCC11778).	(Al Khateeb <i>et al.</i> , 2012)
Ethanollic extract	Hot methanolic extract at 60°C was potent against <i>Pseudomonas aeruginosa</i> (ATCC 27853), <i>Enterobacter aerogenes</i> (ATCC 13048), <i>Klebsiella pneumoniae</i> (ATCC 10031) and <i>Bacillus subtilis</i> (ATCC 6633). Ethanollic extract showed antibacterial activity against <i>Staphylococcus aureus</i> (ATCC 29213), <i>Bacillus cereus</i> (ATCC11778) and <i>Klebsiella pneumoniae</i> (ATCC 10031) Of both <i>ex-vitro</i> and <i>in-vitro</i> growing plantlets and callus cultures.	
Methanolic seeds extract	Methanolic seeds extract showed high antimicrobial activity against Gram-positive bacteria: <i>Staphylococcus aureus</i> (ATCC 25923) and <i>Bacillus cereus</i> (ATCC 33018), Gram-negative bacteria: <i>Salmonella typhimurium</i> (NCTC 12023/ATCC 14028) and <i>Escherichia coli</i> ATCC 25922) and the fungi: <i>Candida albicans</i> (CAIM-22).	(Amer, 2018)
Methanolic leaves extract	Methanolic leaves and roots extract showed a high antibacterial activity against Gram-positive bacteria: <i>Staphylococcus aureus</i> (ATCC 25923) and <i>Bacillus cereus</i> (ATCC 33018).	
Methanolic roots extract		
Water, ethanollic, methanolic and acetone extracts	Marked antifungal activity against <i>Aspergillus aflatoxiformans</i> and <i>Aspergillus ochraceous</i> while the methanolic extract is the more potent.	(Mostafa and El-Sayed, 2021)
III- Antioxidant activity		
Ethanollic extract	Significant antioxidant activity by inhibition of intracellular reactive oxygen species (ROS) production thus protection against hepatic damage induced by tert-butyl hydroperoxide (<i>t</i> -BHP)- in HepG2 cell line <i>in-vitro</i> .	(Chen <i>et al.</i> , 2011)
Ethanollic extract	Potent antioxidant and anti-lipid peroxidative effects	(Chen <i>et al.</i> , 2011)

	shown by remarkable reduction in malondialdehyde (MDA) level in the mice's liver tissue treated with <i>t</i> -BHP <i>in-vivo</i> .	
Kaempferol Kaempferol-3-O-β-D-glucoside Adenosine and 2-furanmethanol-(5'→11)-1,3-cyclopentadiene-[5,4-c]-1H-cinnoline	Potent antioxidant activity of kaempferol followed by kaempferol-3-O-β-D-glucoside Moderate antioxidant activity of adenosine and 2-furanmethanol-(5'→11)-1,3-cyclopentadiene-[5,4-c]-1H-cinnoline in Oxygen Radical Absorbance Capacity (ORAC) Assay.	(Chen <i>et al.</i> , 2011)
Aqueous suspension of leaves powder	Significant decrease in liver superoxide dismutase (SOD) activity, and significant increase in antioxidant enzymes; glutathione-S-transferase (GST) activity and glutathione (GSH) levels <i>in-vivo</i> in Streptozotocin (STZ)-induced diabetic rats.	(Kamel and Marzouk, 2011)
Methanolic extract	The methanolic extract growing <i>ex-vitro</i> plantlets and callus cultures show significant antioxidant potential due to high total phenol content.	(Al Khateeb <i>et al.</i> , 2012)
Aesculetin Quercetin Astragalin Caffeic acid	Marked <i>in-vitro</i> DPPH radical scavenging activities.	(Khalil and Kamel, 2015)
IV-Anti-inflammatory activity		
Ethyl acetate root extract	Significant reduction serum level of IL-6 both in acutely inflamed mice <i>in-vivo</i> caused by lipopolysaccharide LPS and in rats with liver fibrosis.	(Han <i>et al.</i> , 2021)
Ethyl acetate root extract	Significant improve in the damage of colon tissue, reduction of the inflammation of colon, improvement of the lesions in the colonic tissue and reducing necrosis <i>in-vivo</i> in rats with colitis caused by 2,4,6-trinitrobenzenesulphonic acid (TNBS) -ethanol enemas.	(Han <i>et al.</i> , 2021)
Lactucin	Significant reduction in levels of inflammatory mediators' production (IL-6, nitric oxide NO). Significant inhibition of mRNA expression of genes responsible for production of inflammatory mediators (iNOS, COX-2, IL-6, IL-1β). Significant inhibition of Protein Expression iNOS, COX-2 induced by lipopolysaccharide LPS- in RAW264.7 Cells <i>in-vitro</i> .	(Han <i>et al.</i> , 2021)

	<p>Significant inhibition of the phosphorylation and activation of the MAPK-AKT signaling pathway in RAW264.7 cells induced by lipopolysaccharide LPS <i>in-vitro</i>.</p> <p>Suppress the phosphorylation of ERK1/2 and p38 signaling pathways, leading to the inhibition of NO production in RAW 264.7 cells.</p>	
V-Hepatoprotective activity		
Root extract	<p>Normalize some morpho-functional liver features <i>in-vivo</i> in rats with hepatitis induced by CCl₄ as it can decrease cell of necrosis, glycogen content and increase the number of cells with remarkable protein synthesis activity.</p>	(Krylova et al., 2006; Masoud et al., 2018)
Ethanollic extract	<p>Marked reduction in (<i>t</i>-BHP)-induced cell death thus protection against hepatic damage in HepG2 cell line <i>in-vitro</i>.</p>	(Chen et al., 2011)
Ethanollic extract	<p>Significant reduction in serum levels of ALT and AST in <i>t</i>-BHP-induced acute liver injury <i>in-vivo</i> in mice model thus protection against hepatic tissue damage.</p>	(Chen et al., 2011)
Ethyl acetate root extract	<p>Significant improve of liver congestion and normalization of the color of liver.</p> <p>Significant improve in the histopathological changes including reduction of the collagen fibrillar content in liver fibrosis in rats.</p> <p>Significant reduction of serum levels of AST and γ-GT in liver injury-hepatic fibrosis <i>in-vivo</i> model in rats caused by 2,4,6-trinitrobenzenesulphonic acid (TNBS) -ethanol enemas.</p> <p>Can improve <i>in-vitro</i> the impaired intestinal microbes in the gut of rats with liver fibrosis making the intestinal flora close to the normal level.</p> <p>Significant improve of liver fibrosis through the “gut-liver axis” via decreasing intestinal inflammation and promoting probiotic growth (<i>Bifidobacterium Adolescentis</i>).</p>	(Han et al., 2021)
VI-Antidiabetic activity		
Aqueous suspension of leaves powder	<p>Significant decrease in serum levels of the enzymes: aminotransferases (AST, ALT), alkaline phosphatase (ALP), and lactate dehydrogenase (LDH) activities <i>in-vivo</i> in Streptozotocin (STZ)-induced diabetic rats similar to the diabetic drug (glibenclamide) effects.</p>	(Kamel and Marzouk, 2011)

VII-Analgesic and sedative activities		
Lactucin lactucopicrin 11,13-dihydrolylactucin	Marked analgesic and sedative activities in thermal models <i>in-vivo</i> in mice.	(Wesolowska <i>et al.</i> , 2006)
VIII-Skin protective activity		
Ethanollic root extract	Can absorb radiation in the UVB spectrum and prevent <i>in-vivo</i> UVB-induced erythema of human skin. Can <i>in-vitro</i> prevent pyrimidine dimer formation, cell death, and IL-6 mRNA expression in a human keratinocyte cell line after UVB irradiation, so that it can be useful as sunscreen.	(Enk <i>et al.</i> , 2004)
IX-Effect on germination and seedling growth of maize		
Luteolin	Enhances germination and seedling growth of maize in normal conditions. Eliminate the harmful effect of salinity on seedling growth and germination of maize. Stimulates α -amylase activity thus improves mobilization of starch and enhances the accumulation of soluble sugars. Partially enhances the antioxidative defense.	(El-Shafey and AbdElgawad, 2012)
X-Antihypertensive activity		
Methanolic extract	<i>Ex-vitro</i> growing plants and growing callus showed high Angiotensin Converting Enzyme (ACE) inhibitor activity which can be a useful therapy for hypertensive patients by controlling blood pressure.	(Al Khateeb <i>et al.</i> , 2012)

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